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Azoles as Suzuki Cross-Coupling Leaving Groups: Syntheses of 6-Arylpurine 2'-Deoxynucleosides and Nucleosides from 6-(Imidazol-1-yl)- and 6-(1,2,4-Triazol-4-yl)purine Derivatives¹

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ABSTRAC1

6-(Imidazol-1-yl)-, 6-(benzimidazol-1-yl)-, and 6-(1,2,4-triazol-4-yl)purine nucleosides undergo a nickel-mediated C-C cross-coupling of azole-substituted purine derivatives with arylboronic acids to give good yields of 6-arylpurine nucleosides.

Modified purines and purine nucleoside derivatives play a major role in biology, biochemistry, and pharmaceutics.² Recently, 6-arylpurine ribonucleosides have been shown to possess cytostatic activity.³ Classic methodology for the synthesis of biaryls by the Suzuki-Miyaura protocol involves the Pd/Ni-mediated cross-coupling of haloaromatic or arylsulfonate derivatives with arylboronic acids.^{3,4}

We recently demonstrated that 6-iodopurine nucleoside derivatives are markedly superior to their 6-chloropurine analogues as substrates for the Suzuki-Miyaura and other cross-coupling procedures.¹ However, syntheses of such 6-halopurine 2'-deoxynucleosides from naturally occurring 2'-deoxy(inosine/adenosine) are problematic, and high yields are obtained only with considerable care and persistence.^{1,5} By contrast, 6-(imidazol-1-yl)purine (2'-deoxy)nucleoside derivatives can be prepared readily in excellent yields from (2'-deoxy)inosine.⁶ Because of these considerations, we have probed the unexplored utility of 6-(imidazol-1-yl)purine nucleosides as substrates for cross-coupling with arylboronic acids. Such couplings would provide a new avenue for modifications at C6 of purine nucleosides from readily accessible 6-azolyl precursors.

Our initial attempts to couple 4-methoxyphenylboronic acid and 6-(imidazol-1-yl)purine nucleoside derivatives with palladium-based catalyst systems were not successful. Hartwig

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recently prepared three-coordinate arylpalladium amido complexes, which were subjected to reductive elimination to give *N*-coupled arylamines. However, the precursor three-coordinate arylpalladium amido complexes were prepared by treatment of an arylpalladium bromide complex with the potassium salt of a diarylamine. This is drastically different from palladium insertion into the C–N bond of a 6-(imidazol-1-yl)purine. Nickel catalysts have been used successfully in a wide variety of Suzuki reactions, which provide ample precedent for transmetalations with arylboronic acids. Our challenge was to identify a catalytic complex that could insert readily into the purine—imidazole (C6–N) bond.

Imidazolium carbene ligands (Figure 1) have been used successfully in various cross-coupling reactions. On the basis

Figure 1. Structures of the imidazolium carbene ligands IPr (1) and SIPr (2).

of the studies of Blakey and MacMillan, ¹⁰ we examined Ni-(COD)₂ as a catalyst with addition of IPr•HCl for the crosscoupling of 6-(imidazol-1-yl)-9-[2,3,5-tri-O-(4-methylbenzoyl)- β -D-ribofuranosyl]purine (3) and 4-methoxyphenylboronic acid (4) (Scheme 1). However, none of the coupling product,

Scheme 1. Model Coupling Reaction

6-(4-methoxyphenyl)-9-[2,3,5-tri-O-(4-methylbenzoyl)- β -D-ribofuranosyl]purine (5), was detected (Table 1, entry 9). Next, we investigated the catalyst complex resulting from Ni(COD)₂ and SIPr•HCl in the presence of K₃PO₄. We were delighted that the cross-coupled adduct 5 was produced in

Table 1. Cross-Coupling Reaction Conditions and Yields^a

entry	catalyst	ligand	base	5 (%)
1	Pd(PPh ₃) ₄	none	K ₂ CO ₃	< 5
2	Pd(PPh ₃) ₄	SIPr	K_3PO_4	< 5
3	Pd(OAc) ₂	SIPr	K_3PO_4	< 5
4^{b}	Ni(dppp)Cl ₂	SIPr	K_3PO_4	30
5	$Ni(COD)_2$	SIPr	KF	< 5
6	$Ni(COD)_2$	SIPr	CsF	64
7	$Ni(COD)_2$	SIPr	K_3PO_4	83
8^c	$Ni(COD)_2$	SIPr	K_3PO_4	< 5
9	$Ni(COD)_2$	IPr	K_3PO_4	< 5

^a Reaction conditions: 1.0 equiv of **3**, 2.0 equiv of **4**, 10 mol % of catalyst, 10 mol % of ligand, 3.0 equiv of base, 60 °C, 8 h. ^b BuLi (0.4 equiv) was used to reduce Ni(II) to Ni(0). ^c Ambient temperature instead of 60 °C.

high yield (83% isolated) (entry 7). Heating (60 °C) was required to achieve reasonable reaction rates (entry 8). Replacement of Ni(COD)₂ by palladium catalysts in analogous coupling reaction mixtures did not give coupling products in meaningful yields.

The superior reaction efficiency observed with the Ni(0)· SIPr system and K_3PO_4 as base prompted additional evaluation with this catalytic combination. Potential electronic effects on the coupling of 6-(imidazol-1-yl)purine nucleosides by the aryl substituent of the boronic acids was then investigated. Both electron-rich and electron-poor arylboronic acids underwent coupling with 3 in good yields (Schemes 1 and 2) (Table 1, entry 7; Table 2, entries 1–3).

Scheme 2. Coupling Reactions with Varied Substrates

This methodology is efficient for conversions of inosine into various 6-arylpurine ribonucleosides, but alternative cross-coupling reactions with 6-halopurine nucleosides provide comparatively convenient approaches. However, syntheses of 6-halopurine 2'-deoxynucleosides are more chal-

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Table 2. Yields of Coupling Products with Varied Substrates

entry	R_1	R_2	yield (%)
1	OTol	Н	73
2	OTol	CH_3	81
3	OTol	F	78
4	Н	Н	68
5	Н	CH_3	61
6	Н	OCH_3	75
7	Н	F	65

lenging because of the markedly less stable glycosyl linkage of the 2'-deoxy analogues, which can result in cleavage under even mildly acidic conditions. 1.5 Our modified Appel methodology provides convenient conversions of 2'-deoxyinosine derivatives into 6-(imidazol-1-yl)purine 2'-deoxynucleoside analogues in excellent yields (>90%) with virtually no glycosyl bond cleavage. Application of the present coupling protocol to such protected 2'-deoxynucleosides gave the corresponding 6-arylpurine products in good isolated yields (Table 2, entries 4–7).

Sensitivity to the azole substituent was probed with the 6-(benzimidazol-1-yl)purine nucleoside derivative 6 (Figure 2) (also prepared in excellent yield by our modified Appel

Figure 2. Nucleoside derivative 6.

procedure⁶). Coupling of **6** (under the noted conditions) gave **5** (80% isolated yield), which demonstrated that azoles other than imidazole could be used.

Our modified-Appel approach for quantitative conversion of 6-oxopurine (2'-deoxy)nucleoside derivatives into the

Scheme 3. Couplings with 6-(1,2,4-Triazol-4-yl)purine 7

corresponding 6-(azolyl)purine analogues⁶ is buttressed by our methodology for elaboration of the amino group of 6-aminopurine (2'-deoxy)nucleosides into their 6-(1,2,4-triazol-4-yl)purine counterparts.¹¹ Thus, such 6-(azolyl)purine (2'-deoxy)nucleosides are readily available by convenient transformations of the natural products (2'-deoxy)adenosine¹¹ and (2'-deoxy)inosine,⁶ as well as for other naturally occurring and synthetic analogues. A brief investigation of the 6-(1,2,4-triazol-4-yl)purine system was then undertaken (Scheme 3, Table 3).

Table 3. Yields of **8** with 6-(1,2,4-Triazol-4-yl)purine **7**

entry	R	8 (%)
1	Н	80
2	CH_3	78
3	$ m CH_3$ $ m OCH_3$	85
4	F	75

The noted reaction conditions were applied to a cross-coupling of 9-(3,5-di-O-acetyl-2-deoxy- β -D-erythro-pento-furanosyl)-6-(1,2,4-triazol-4-yl)purine¹¹ (7) and 4-methoxy-phenylboronic acid (4). Two products, **8c/9c** (45:55), were obtained. Fortunately, the use of IPr•HCl as ligand and CsF as base gave the desired cross-coupling product **8c** in high yield (85%), with the oxygen-insertion byproduct **9c** detected as a minor impurity (\sim 5%). This coupling also was found to be tolerant with respect to substituents on the arylboronic acid component [with trace oxygen-insertion byproduct formation (\leq 5%)].

In summary, we now report nickel-based systems with imidazolium carbene ligands, which catalyze efficient Suzuki cross-coupling of arylboronic acids and 6-(imidazol-1-yl)-, 6-(benzimidazol-1-yl)-, and 6-(1,2,4-triazol-4-yl)purine (2'-deoxy)nucleoside derivatives to provide the corresponding 6-arylpurine (2'-deoxy)nucleosides. Different ligand/base combinations give better results with imidazole versus triazole substrates. Novel 6-(aryloxy)purine 2'-deoxynucleosides, oxygen-insertion byproducts, were observed with the 6-(1,2,4-triazol-1-yl)purine derivatives. Further studies are in progress with different substrates and catalytic systems.

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Supporting Information Available: Experimental details, characterization data, and ¹H NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

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